

Go Direct[®] Chloride Ion- Selective Electrode (Order Code GDX-CL)



Go Direct Chloride Ion-Selective Electrode (ISE) is used to measure the concentration of chloride (Cl^-) ions in aqueous samples. It is designed to be used with the Vernier Go Direct ISE Amplifier (order code GDX-ISEA). Chloride ions are found in freshwater samples as a result of water flowing over salt-containing minerals. These salts might include either sodium chloride (NaCl) or potassium chloride (KCl). The EPA maximum contamination level for chloride concentration in drinking water is 250 mg/L. The chloride ion concentration in seawater is approximately 19,400 mg/L—well below the upper limit of the Chloride ISE of 35,000 mg/L.

Note: Vernier products are designed for educational use. Our products are not designed nor are they recommended for any industrial, medical, or commercial process such as life support, patient diagnosis, control of a manufacturing process, or industrial testing of any kind.

What's Included

- Go Direct Chloride Ion-Selective Electrode (Go Direct ISE Amplifier connected to a Go Direct Chloride Ion-Selective Electrode BNC)
- 30 mL bottle of High Standard solution with SDS (1000 mg/L Cl^-)
- 30 mL bottle of Low Standard solution with SDS (10 mg/L Cl^-)
- Short-Term ISE Soaking Bottle
- Polishing strip
- Micro USB Cable

Compatible Software

See www.vernier.com/manuals/gdx-cl for a list of software compatible with the Go Direct Chloride Ion-Selective Electrode.

Getting Started

Please see the following link for platform-specific connection information:

www.vernier.com/start/gdx-cl

Bluetooth Connection

1. Install Graphical Analysis 4 on your computer, Chromebook[™], or mobile device. See www.vernier.com/ga4 for software availability.
2. Charge your sensor for at least 2 hours before first use.

USB Connection

1. Prepare the electrode by soaking it in the High Standard solution for 30 minutes. Refer to the Using the Product section for more information.
2. Install Graphical Analysis 4 on your

3. Prepare the electrode by soaking it in the High Standard solution for 30 minutes. Refer to the Using the Product section for more information.
4. Turn on your sensor by pressing the power button once. The Bluetooth[®] LED will blink red.
5. Launch Graphical Analysis 4.
6. Click or tap Sensor Data Collection.
7. Click or tap your Go Direct sensor from the list of Discovered Wireless Devices. Your sensor's ID is located near the barcode on the sensor. The Bluetooth LED will blink green when it is successfully connected.
8. This is a multi-channel sensor. The active channel is listed in the Connected Devices Sensor Channels list. To change channels, select the check box next to the Sensor Channel(s) you would like to activate.
9. Click or tap Done to enter data-collection mode.
10. For best results, perform a two-point calibration using the High and Low Standard solutions.

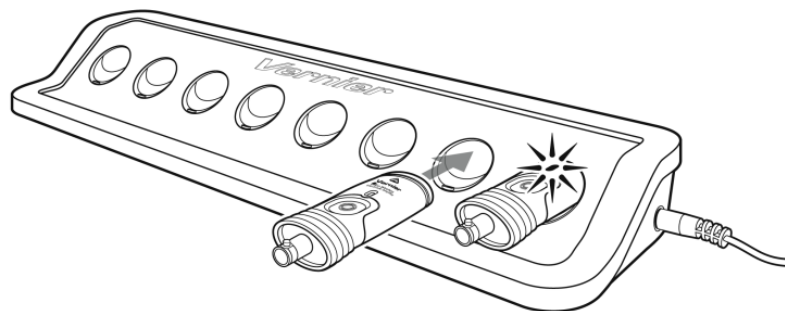
computer or Chromebook. If using LabQuest 2, make sure LabQuest App is up to date. See www.vernier.com/ga4 for software availability or www.vernier.com/downloads to update LabQuest App.

3. Connect the sensor to the USB port.
4. Launch Graphical Analysis 4 or turn on LabQuest 2.
5. This is a multi-channel sensor. To change the default channel selections, see www.vernier.com/start/gdx-cl.
6. For best results, perform a two-point calibration using the High and Low Standard solutions.

Charging the Sensor

Connect the Go Direct Chloride Ion-Selective Electrode to the included Micro USB Cable and any USB device for two hours. Connecting the Go Direct Chloride BNC Electrode to the amplifier during charging is optional.

You can also charge up to eight Go Direct Chloride Ion-Selective Electrodes using our Go Direct Charge Station, sold separately (order code: GDX-CRG). An LED on each Go Direct Chloride Ion-Selective Electrode indicates charging status.



Charging	Blue LED on steady while sensor is connected to the Micro USB Cable or Charge Station.
Fully charged	Blue LED is off when charging is complete.

Powering the Sensor

Turning on the sensor	Press button once. Red LED indicator flashes when unit is on.
Putting the sensor in sleep mode	Press and hold button for more than three seconds to put into sleep mode. Red LED indicator stops flashing when sleeping.

Connecting the Sensor

See the following link for up-to-date connection information:

www.vernier.com/start/gdx-cl

Connected and charging	Blue and Green LED solid when sensor is connected to Graphical Analysis via USB and unit is charging. (Green LED is obscured by the blue one.)
Connected, fully charged	Green LED solid when sensor is connected to Graphical Analysis via USB and the unit is fully charged.
Charging via USB, connected via Bluetooth	Blue LED is solid and green LED is flashing, but the green flashing LED looks white because it is overwhelmed by the blue.

Identifying the Sensor

When two or more sensors are connected, the sensors can be identified by

tapping or clicking Identify in Sensor Information.

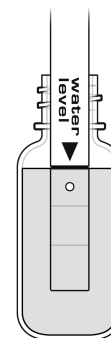
Using the Product

1. Remove the storage bottle from the electrode by unscrewing the lid and removing the bottle and lid.
2. Thoroughly rinse the lower section of the probe using distilled or deionized water.
3. Soak the tip of the electrode for 30 minutes in the High Standard solution.
 - The ISE should not rest on the bottom of the container.
 - The small white reference contacts near the tip of the electrode should be immersed.
 - Make sure no air bubbles are trapped below the ISE.
4. Connect the sensor following the steps in the Getting Started section.
5. For best results, perform a two-point calibration using the High and Low Standard solutions. For calibration instructions, see www.vernier.com/tit/4011
6. When you are finished making measurements, rinse the electrode with distilled water.
7. Slide the cap onto the electrode body, and then screw the cap onto the storage bottle so the tip of the electrode is not touching the sponge.

Important: Do not fully submerge the sensor. The BNC connection is not waterproof.

Important: Do not leave the ISE soaking for more than 24 hours.

Note: If the ISE needs to be transported to the field during the soaking process, use the Short-Term ISE Soaking Bottle. Remove the cap from the bottle and fill it 3/4 full with High Standard. Slide the bottle's cap onto the ISE, insert it into the bottle, and tighten it. For long-term storage, greater than 24 hours, make sure the sensor is stored in its storage bottle with the sponge slightly damp.



Channels

Go Direct Chloride Ion-Selective Electrode has six sensor channels. The channel names are

- Potential (mV)
- Chloride (mg/L)

- Ammonium (mg/L)
- Calcium (mg/L)
- Nitrate (mg/L)
- Potassium (mg/L)

Note: The Chloride channel is the default channel for this sensor. All channels are mutually exclusive except Potential (i.e., You can display one concentration channel and Potential at the same time, but you cannot display two concentration channels at the same time). In order to collect data from the other concentration channels, you must also attach the applicable corresponding BNC electrode to the amplifier.

Calibrating the Sensor

A calibration is stored on each sensor before it is shipped. As the membrane ages, this factory calibration may become inadequate. For best results, we recommend performing a two-point calibration.

Note: If you plan to use the electrode outside the range of the standards provided, you will need to prepare your own standards and use those for soaking and calibration. Standards should be two decades apart (e.g., 5 mg/L and 500 mg/L).

For additional calibration information, see www.vernier.com/til/4011

Specifications

Range (concentration)	1 to 35,000 mg/L (or ppm)
Accuracy after calibration	±10% of full scale (calibrated 10 to 1000 mg/L)
Interfering ions	CN ⁻ , Br ⁻ , I ⁻ , OH ⁻ , S ²⁻ , NH ₃
pH range	2–12 (no pH compensation)
Temperature range	0–80°C (no temperature compensation)
Electrode slope	–56 ±3 mV/decade at 25°C
Electrode resistance	1 to 5 MΩ
Minimum sample size	Must be submerged 2.8 cm (1.1 in)
Potential readings, typical	114 mV in 1000 mg/L, 230 mV in 10 mg/L
USB specification	2.0
Wireless specification	Bluetooth 4.2
Maximum wireless range	30 m
Battery	300 mA Li-Poly

Battery life (single full charge)	~24 hours
Battery life (long term)	~500 full charge cycles (several years depending on usage)

Care and Maintenance

Proper care and storage are important for optimal longevity of your Chloride ISE.

- Long-term storage of the ISE (longer than 24 hours): Moisten the sponge in the bottom of the long-term storage bottle with distilled water. When you finish using the ISE, rinse it off with distilled water and blot it dry with a paper towel. Loosen the lid of the long-term storage bottle and insert the ISE. **Note:** The tip of the ISE should NOT touch the sponge. Also, make sure the white reference mark is inside the bottle. Tighten the lid. This will keep the electrode in a humid environment, which prevents the reference junctions from completely drying out.
- Put the device in sleep mode by holding the button down for at least three seconds. The red LED will stop flashing to show that the unit is in sleep mode. Over several months, the battery will discharge but will not be damaged. After such storage, charge the device for a few hours, and the unit will be ready to go.
- Short-term wet storage (less than 24 hours): Fill the Short-Term ISE Soaking Bottle 3/4 full with High Standard. Loosen the cap, insert the electrode into the bottle, and tighten it.

Note: Exposing the battery to temperatures over 35°C (95°F) will reduce its lifespan. If possible, store the device in an area that is not exposed to temperature extremes.

Maintaining and Replacing the ISE Standard Calibration Solutions

Having accurate standard solutions is essential for performing good calibrations. The two standard solutions that were included with your ISE can last a long time if you take care not to contaminate them. At some point, you will need to replenish your supply of standard solutions. Vernier sells 500 mL bottles of Chloride Low Standard, 10 mg/L (order code CL-LST) and Chloride High Standard, 1000 mg/L (order code CL-HST).

To prepare your own standard solutions, use the information in the following table. **Note:** Use glassware designed for accurate volume measurements, such as volumetric flasks or graduated cylinders. All glassware must be very clean.

Standard Solution	Concentration (mg/L or ppm)	Preparation Method using High-Quality Distilled Water
Chloride (Cl ⁻) ISE High Standard	1000 mg/L as Cl	1.648 g NaCl/ 1 L solution
Calcium (Cl ⁻) ISE Low Standard	10 mg/L as Cl	Dilute the High Standard by a factor of 100 (from 1000 mg/L to 10 mg/L).*

*Perform two serial dilutions as described below.

- Combine 100 mL of the High Standard with 900 mL of distilled water. Mix well.
- Combine 100 mL of the solution made in the previous step with 900 mL of distilled water. Mix well.

Battery Information

The Go Direct Chloride Ion-Selective Electrode contains a small lithium-ion battery in the handle. The system is designed to consume very little power and not put heavy demands on the battery. Although the battery is warranted for one year, the expected battery life should be several years. Replacement batteries are available from Vernier (order code: GDX-BAT-300).

Water Resistance

The Go Direct Chloride Ion-Selective Electrode is not water resistant and should never be immersed in water above the BNC junction.

If water gets into the device, immediately power the unit down (press and hold the power button for more than three seconds). Disconnect the sensor and charging cable, and remove the battery. Allow the device to dry thoroughly before attempting to use the device again. Do not attempt to dry using an external heat source.

How the Sensor Works

Combination Ion-Selective Electrodes consist of an ion-specific (sensing) half-cell and a reference half-cell. The ion-specific half-cell produces a potential that is measured against the reference half-cell depending on the activity of the target ion in the measured sample. The ion activity and the potential reading change as the target ion concentration of the sample changes. The relationship between the potential measured with the ISE and the ion activity, and thereby the ion concentration in the sample, is described by the Nernst equation:

$$E = E_o - 2.303 \frac{RT}{nF} \log(C + C_o)$$

E = measured potential (mV) between the ion-selective and the reference electrode

E_o = standard potential (mV) between the ion-selective and reference electrodes

R = universal gas constant ($R = 8.314 \text{ J mol}^{-1} \text{ K}^{-1}$)

T = temperature in K (Kelvin), with $T (\text{K}) = 273.15 + t \text{ } ^\circ\text{C}$ where t is the temperature of the measured solution in $^\circ\text{C}$.

F = Faraday constant (96485 C mol^{-1})

n = valence of the ion

C = concentration of ion to be measured

C_o = detection limit

Since R and F are constant, they will not change. The electrical charge of the ion (valence) to be measured is also known. Therefore, this equation can be simplified as:

$$E = E_o - S \cdot \log(C + C_o)$$

where $S = -2.303 \frac{RT}{nF}$ is the ideal slope of the ISE.

The following table describes ideal behavior:

Ion Examples	n (valence of ion)	S (at 25 °C), mV/decade
Calcium (Ca ²⁺)	+2	+29.58
Potassium (K ⁺), Ammonium (NH ₄ ⁺)	+1	+59.16
Nitrate (NO ₃ ⁻), Chloride (Cl ⁻)	-1	-59.16

Assuming C_o is near zero, the equation can be rewritten as:

$$C = 10^{[(E - E_o) / S]}$$

allowing for the calculation of the ion concentration.

It is very important to note that this table reflects ideal behavior. Ion-selective electrodes have slopes that are typically lower than ideal. It is generally accepted that an ISE slope from 88–101% of ideal is allowable. The slope (S) is an indicator of ISE performance. If the slope changes significantly over time, it may indicate that it is necessary to replace the ISE sensor tip.

Potential vs. Concentration

To measure the mV readings from an aqueous sample, calibration is not required. To convert mV readings to concentration (mg/L or ppm), the software uses a modified version of the Nernst Equation:

$$C = 10^{[(E - E_o) / S_m]}$$

C = concentration of ion to be measured (mg/L or ppm)

E = measured potential of sample (mV)

E_o = measured potential (mV) at a $C = 1 \text{ mg/L Cl}^-$ concentration

S_m = measured electrode slope in mV/decade

The value of S_m , the measured electrode slope, is determined by measuring the potential of two standard solutions, and solving the equation below:

$$S_m = -[(\text{Low Standard} - \text{High Standard}) / \# \text{ of decades}^*]$$

*A decade is defined as the factor of the difference between the two standard solutions. For example, the difference between a 1mg/L standard and a 100 mg/L standard is 2 decades (a factor of 100 difference, or 1×10^2).

Example Calculation, converting mV to mg/L

For this example, the measured quantities are shown in the chart below:

Solution	Measured Potential
1 mg/L Cl ⁻ standard	288 mV
10 mg/L Cl ⁻ standard	230 mV
1000 mg/L Cl ⁻ standard	114 mV
unknown sample	188 mV

$$S_m = -\frac{(230 \text{ mV} - 114 \text{ mV})}{2 \text{ decades}} = -58 \text{ mV/decade}$$

$$C = 10^{[(188 \text{ mV} - 288 \text{ mV}) / -58 \text{ mV/decade}]} = 53 \text{ mg/L Cl}^-$$

Troubleshooting

Sampling Freshwater Samples for Chloride Concentration

For best results, calibrate the Chloride ISE using the 10 mg/L and 1000 mg/L standards.

Measuring Chloride Concentration of Saltwater or Brackish Water

When measuring chloride concentration in seawater or brackish water, calibrate the Chloride ISE using the 1000 mg/L standard included with your Chloride ISE for one calibration point (or 1.806 parts per thousand, or ppt). For the second calibration point, prepare a standard that is 20,000 mg/L Cl⁻ by adding 32.96 g of solid NaCl to enough distilled water to prepare 1 L of solution:

$$\frac{2000 \text{ mg Cl}^-}{1 \text{ L}} \times \frac{1 \text{ g Cl}^-}{1000 \text{ mg Cl}^-} \times \frac{58 \text{ g NaCl}}{355 \text{ g Cl}^-} = 32.96 \text{ g NaCl/L solution}$$

If you are calibrating in ppt, call this solution 36.13 ppt.

Determining Salinity of Saltwater or Brackish Water

Salinity is the total of all salts dissolved in water, expressed either as mg/L (equal to parts per million, ppm) or in parts per thousand (ppt). Seawater contains a fairly constant quantity of chloride ions. From your measurement of chloride ion concentration (in the previous section), salinity can be calculated using the following formula:

$$\text{Salinity (mg/L or ppm)} = 1.8066 \times [\text{Cl}^- \text{ concentration, mg/L}]$$

Using this formula, the salinity of saltwater is calculated to be:

$$\text{Salinity (mg/L or ppm)} = 1.8066 \times (19400 \text{ mg/L}) = 35,000 \text{ mg/L}$$

The level of salinity of seawater in parts per thousand, or ppt, would be:

$$\text{Salinity (ppt)} = 35000 / 1000 = 35 \text{ ppt}$$

Using Ionic Strength Adjuster Solutions to Improve Accuracy

For optimal results at low concentrations of chloride ions, a standard method for taking measurements with the Chloride Ion-Selective Electrode (ISE) is to add ionic strength adjuster (ISA) solutions to each of your standard solutions and samples.

Adding an ISA ensures that the total ion activity in each solution being measured is nearly equal, regardless of the specific ion concentration. This is especially important when measuring very low concentrations of specific ions. The ISA contains no ions common to the Chloride ISE itself. **Note:** The additions of ISA to samples or standards described below do not need to have a high level of accuracy—combining the ISA solution and sample solution counting drops using a disposable Beral pipet works fine.

Use an ISA with the Chloride ISE by adding 5.0 M NaNO₃ ISA solution (42.50 g NaNO₃ / 100 mL solution) to the Cl⁻ standard or to the solution being measured, in a ratio of 1 part of ISA (by volume) to 50 parts of the total solution (e.g., 1 mL of ISA to 50 mL of total solution, or 2 drops of ISA to 5 mL of total solution).

When the response of the Chloride ISE begins to slow, the membrane may need polishing. Cut a small piece (about 1 inch square) from a polishing strip. Wet the end of the electrode and the dull side of the polishing strip thoroughly with distilled water. Using only moderate pressure, polish the end of the electrode by gently rubbing it in a circular motion. This will remove the inactive layer of the membrane which impedes measurement. Rinse thoroughly with distilled water and recalibrate in the usual manner.

See general tips for using Ion Selective Electrodes at www.vernier.com/tit/665

Repair Information

If you have followed the troubleshooting steps and are still having trouble with your Go Direct Chloride Ion-Selective Electrode, contact Vernier Technical Support at support@vernier.com or call 888-837-6437. Support specialists will work with you to determine if the unit needs to be sent in for repair. At that time, a Return Merchandise Authorization (RMA) number will be issued and instructions will be communicated on how to return the unit for repair.

Accessories/Replacements

Item	Order Code
Storage Solution Bottles, pkg of 5	BTL
Standard High CI ISE Solution	CL-HST
Standard Low CI ISE Solution	CL-LST
Go Direct Chloride Ion-Selective Electrode BNC	GDX-CL-BNC
Go Direct ISE Amplifier	GDX-ISEA
Micro USB Cable	CB-USB-MICRO
USB-C to Micro USB Cable	CB-USB-C-MICRO
Go Direct 300 mAh Replacement Battery	GDX-BAT-300

Warranty

Vernier warrants this product to be free from defects in materials and workmanship for a period of five years from the date of shipment to the customer. This warranty does not cover damage to the product caused by abuse or improper use. This warranty covers educational institutions only. ISE modules are covered by a one-year warranty.

Disposal

When disposing of this electronic product, do not treat it as household waste. Its disposal is subject to regulations that vary by country and region. This item should be given to an applicable collection point for the recycling of electrical and electronic equipment. By ensuring that this product is disposed of correctly, you help prevent potential negative consequences on human health or on the environment. The recycling of materials will help to conserve natural resources. For more detailed information about recycling this product, contact your local city office or your disposal service.

Battery recycling information is available at www.call2recycle.org

Do not puncture or expose the battery to excessive heat or flame.



The symbol, shown here, indicates that this product must not be disposed of in a standard waste container.

Federal Communication Commission Interference Statement

This equipment has been tested and found to comply with the limits for a Class B digital device, pursuant to Part 15 of the FCC rules. These limits are designed to provide reasonable protection against harmful interference in a residential installation. This equipment generates, uses and can radiate radio frequency energy and, if not installed and used in accordance with the instructions, may cause harmful interference to radio communications. However, there is no guarantee that interference will not occur in a particular installation. If this equipment does cause harmful interference to radio or television reception, which can be determined by turning the equipment off and on, the user is encouraged to try to correct the interference by one or more of the following measures:

- Reorient or relocate the receiving antenna.
- Increase the separation between the equipment and receiver.
- Connect the equipment into an outlet on a circuit different from that to which the receiver is connected.
- Consult the dealer or an experienced radio/TV technician for help.

FCC Caution

This device complies with Part 15 of the FCC Rules. Operation is subject to the following two conditions:

- (1) this device may not cause harmful interference and
- (2) this device must accept any interference received, including interference that may cause undesired operation

RF Exposure Warning

The equipment complies with RF exposure limits set forth for an uncontrolled environment. The antenna(s) used for this transmitter must not be co-located or operating in conjunction with any other antenna or transmitter. You are cautioned that changes or modifications not expressly approved by the party responsible for compliance could void your authority to operate the equipment.

IC Statement

This device complies with Industry Canada license-exempt RSS standard(s). Operation is subject to the following two conditions:

- (1) this device may not cause interference, and
- (2) this device must accept any interference, including interference that may cause undesired operation of the device.

Industry Canada - Class B This digital apparatus does not exceed the Class B limits for radio noise emissions from digital apparatus as set out in the interference-causing equipment standard entitled "Digital Apparatus," ICES-003 of Industry Canada. Operation is subject to the following two conditions: (1) this device may not cause interference, and

- (2) this device must accept any interference, including interference that may cause undesired operation of the device.

To reduce potential radio interference to other users, the antenna type and its gain should be so chosen that the equivalent isotropically radiated power (e.i.r.p.) is not more than that permitted for successful communication.

RF exposure warning: The equipment complies with RF exposure limits set forth for an uncontrolled environment. The antenna(s) used for this transmitter must not be co-located or operating in conjunction with any other antenna or transmitter.

Le présent appareil est conforme aux CNR d'Industrie Canada applicables aux appareils radio exempts de licence. L'exploitation est autorisée aux deux conditions suivantes :

- (1) l'appareil ne doit pas produire de brouillage, et
- (2) l'appareil doit accepter toute interférence radioélectrique, même si cela résulte à un brouillage susceptible d'en compromettre le fonctionnement.

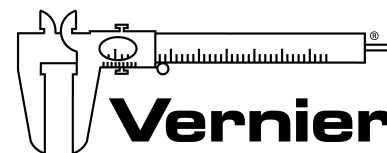
Cet appareil numérique respecte les limites de bruits radioélectriques applicables aux appareils numériques de Classe B prescrites dans la norme sur le matériel interférant-brouilleur: "Appareils Numériques," NMB-003 édictée par Industrie Canada. L'utilisation est soumise aux deux conditions suivantes:

- (1) cet appareil ne peut causer d'interférences, et
- (2) cet appareil doit accepter toutes interférences, y comprises celles susceptibles de provoquer un dysfonctionnement du dispositif.

Afin de réduire les interférences radio potentielles pour les autres utilisateurs, le type d'antenne et son gain doivent être choisis de telle façon que l'équivalent de puissance isotrope émise (e.i.r.p.) n'est pas plus grand que celui permis pour une communication établie.

Avertissement d'exposition RF: L'équipement est conforme aux limites d'exposition aux RF établies pour un environnement non supervisé. L'antenne (s) utilisée pour ce transmetteur ne doit pas être jumelée ou fonctionner en conjonction avec toute autre antenne ou transmetteur.

Note: This product is a sensitive measurement device. For best results, use the cables that were provided. Keep the device away from electromagnetic noise sources, such as microwaves, monitors, electric motors, and appliances.



MEASURE. ANALYZE. LEARN.™

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